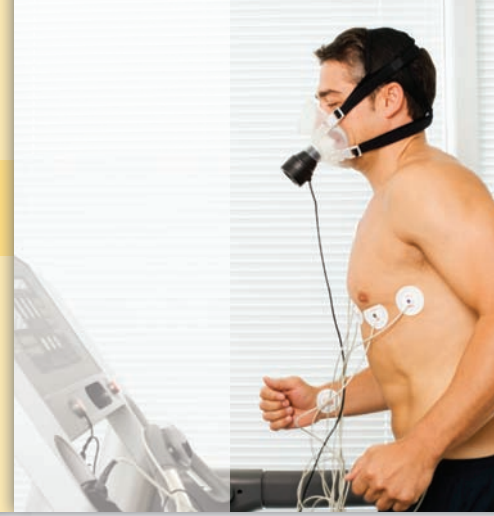


CHAPTER 2

Pulmonary Ventilation



© IT Stock/Polka Dot/Thinkstock

Chapter Objectives

By studying this chapter, you should be able to do the following:

1. Identify the basic structures of the conducting and respiratory zones of the ventilation system.
2. Explain the role of minute ventilation and its relationship to the function of the heart in the production of energy at the tissues.
3. Identify the different ways in which carbon dioxide is transported from the tissues to the lungs.
4. Explain the respiratory advantage of breathing depth versus rate during a treadmill exercise.
5. Describe the composition of ambient air and alveolar air and the pressure changes in the pleural and pulmonary spaces.
6. Diagram the three ways in which carbon dioxide is transported in the venous blood to the lungs.
7. Define pleural pressure. What happens to alveolar volume when pleural pressure decreases?
8. Describe the mechanics of ventilation with respect to the changes in pulmonary pressures.
9. Identify the muscles involved in inspiration and expiration at rest.
10. Describe the partial pressures of oxygen and carbon dioxide in the alveoli, lung capillaries, tissue capillaries, and tissues.
11. Describe how carbon dioxide is transported in the blood.
12. Explain the significance of the oxygen-hemoglobin dissociation curve.
13. Discuss the effects of decreasing pH, increasing temperature, and increasing 2,3-diphosphoglycerate on the HbO₂ dissociation curve.
14. Distinguish between and explain external respiration and internal respiration.

Chapter Outline

Pulmonary Structure and Function

- Anatomy of Ventilation
 - Lungs
- Mechanics of Ventilation
 - Inspiration
 - Expiration
 - Pressure Changes

Pulmonary Volumes and Capacities

- Lung Volumes and Capacities
- Pulmonary Ventilation
 - Minute Ventilation
 - Alveolar Ventilation

Gas Exchange in the Lungs and the Body

Gas Exchange
Ambient Air
In the Trachea
Alveolar Air
In Tissues

Oxygen and Carbon Dioxide Transport

Oxygen Transport in the Blood
In Physical Solution
Combined with Hemoglobin

Oxyhemoglobin (HbO₂) Dissociation Curve
Myoglobin and Storage of Oxygen
Carbon Dioxide Transport in the Blood
In Solution
As Carbamino Compounds
As Bicarbonate
Diffusing Capacity and Transit Time
Ventilation–Perfusion Ratio
Oxygen Delivery and Utilization

Pulmonary ventilation

A term that refers to breathing (ventilation of the lungs).

Cardiorespiratory system

The purpose of the cardiorespiratory (heart and lung) system is to deliver oxygen to the mitochondria of cells that has previously been extracted from the atmosphere via the lungs.

Primary bronchi One of the two main air passages that branch from the trachea and convey air to the lungs as part of the respiratory system.

Secondary bronchi

Secondary bronchi arise from the primary bronchi. Each one serves as the airway to a specific lobe of the lung.

Bronchioles Branches off the bronchi that become increasingly smaller airways that send the air to the alveoli.

Alveoli Tiny saclike dilations where gas exchange (oxygen and carbon dioxide) takes place in the lungs.

Pulmonary Structure and Function

The respiratory system is responsible for the movement of air in and out of the lungs through a process called **pulmonary ventilation**. This system, also known as the ventilation system, meets the body's needs for gas exchange at the lungs and at the tissues. The primary function of pulmonary ventilation is to make oxygen available to the blood, which is transported by the cardiovascular system throughout the body to all the cells. The exchange of oxygen (O₂) and carbon dioxide (CO₂) at the cellular level is necessary to produce energy for muscle contraction while regulating the internal environment of the tissues.

Anatomy of Ventilation

The respiratory and circulatory systems work together to bring in oxygen and transport it to the tissues throughout the body. Together, both systems are known as the **cardiorespiratory system**. This system is primarily responsible for aerobic capacity, which is a significant factor in health and athletics. With an increase in the capacity of the cardiorespiratory system to make more oxygen available at the tissues, more energy can be produced for muscle contraction. This allows the muscles to do more work.

Lungs

Oxygen is moved in and out of the lungs through the conduction zone of the lung tree. This is necessary to bring the atmospheric air temperature to body temperature. The air is also filtered and humidified as it passes through the nose, mouth, and the trachea, which divides into right and left **primary bronchi** that divide further into the **secondary bronchi** within each lung (**Figure 2-1**). The branching tubes become narrower and more numerous as they divide into segmental bronchi, terminal bronchioles, respiratory **bronchioles**, and, finally, alveolar ducts, which are lined with tiny air sacs called **alveoli**.

The **conducting zone** extends from the trachea to the terminal bronchioles. It is composed of 16 generations of tubes branching into smaller tubes supported by a composition of 15–20 C-shaped cartilage rings and smooth muscle to help ensure that the inspired air gets to the respiratory zone. The parasympathetic

division of the autonomic nervous system causes the smooth muscle to constrict, and the sympathetic division causes the smooth muscle to relax (or dilate).

Aside from warming and humidifying the air, the conducting zone is lined with ciliated mucous membranes that filter incoming air to keep the passageways clean. During rest, with each inspiration, about 500 mL of air enters the conducting zone (called **tidal volume**; V_T) to reach the alveoli (the **respiratory zone**), where each alveolus is surrounded by capillaries to permit gas exchange of oxygen and carbon dioxide (Figure 2-2). Because no exchange of gases takes place in the conducting zone, it is called the **anatomical dead space**.

The dead space volume can be estimated as 1 mL of air for each 1 lb of ideal body weight. Thus, a V_T of 500 mL per breath minus the dead space of 150 mL of air for a 150-lb person would approximate the volume of air transported to the alveoli (350 mL). This means 30% of the V_T ($150 \div 500 = 0.3 \times 100 = 30\%$) does not participate in the gas exchange and constitutes wasted air (Table 2-1).

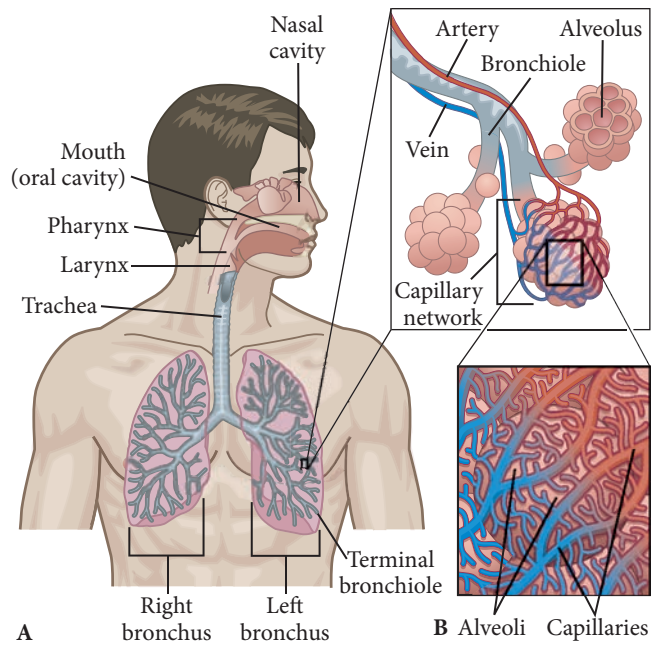


Figure 2-1 The respiratory system. (A) The upper and lower airway divisions. (B) Alveoli.

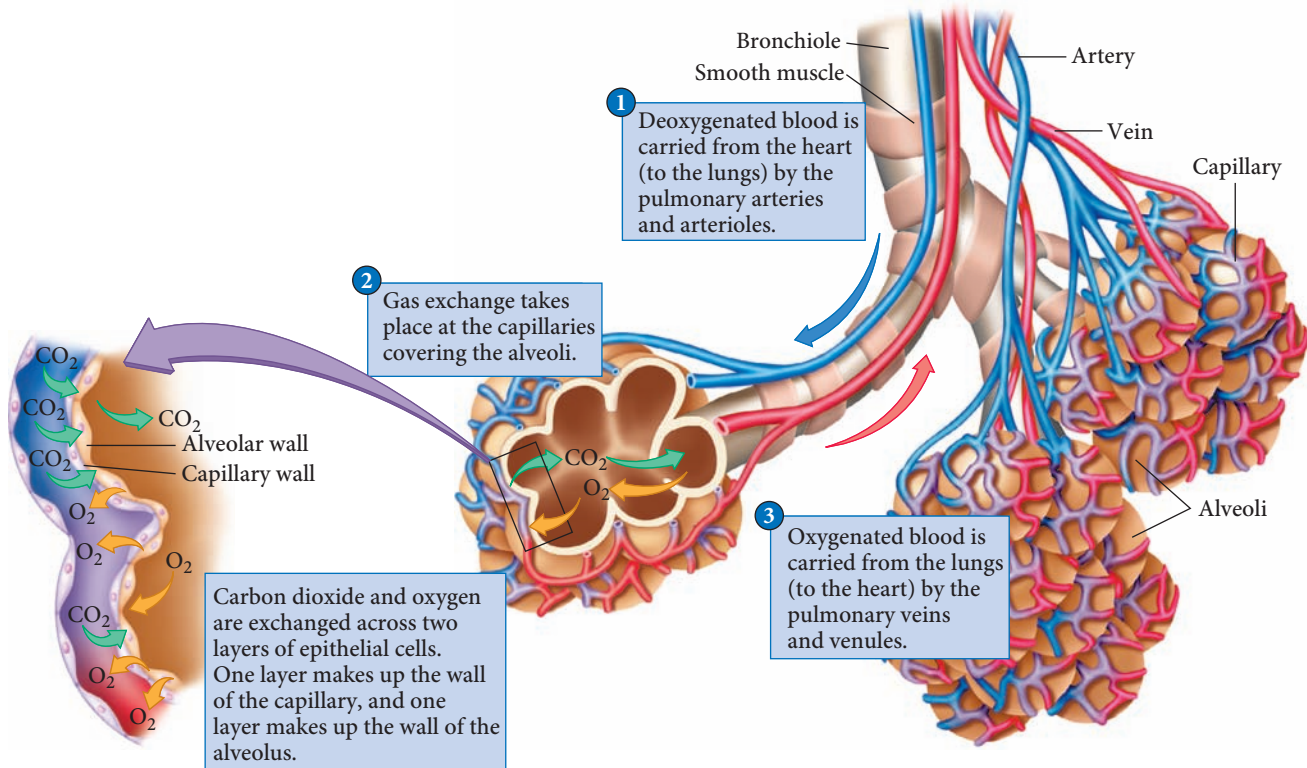


Figure 2-2 Exchange of gases in the lungs.

Conducting zone The nose, pharynx, larynx, trachea, bronchi, bronchioles, and terminal bronchioles function to filter, warm, and moisten air, and conduct it to the alveoli.

Tidal volume The amount of air that is inspired or expired in a normal breathing cycle.

Respiratory zone The respiratory zone is the site of oxygen and carbon dioxide exchange with the blood.

Anatomical dead space All airways such as the mouth, nose, pharynx, larynx, trachea, bronchi, and bronchioles that do not participate directly in the diffusion of oxygen from the alveoli into the pulmonary capillaries.

Gas exchange The diffusion of oxygen and carbon dioxide down their concentration gradients that occurs between the pulmonary capillaries and alveoli and between the systemic capillaries and the tissues.

Hemoglobin The protein portion of the red blood cell that binds with oxygen. It consists of four iron-containing pigments called hemes and a protein called globin.

Oxyhemoglobin When hemoglobin is combined with oxygen, it is referred to as oxyhemoglobin.

Ventilation The movement of air into and out of the lungs.

Barometric (atmospheric) pressure The reading given by a mercury barometer that determines pressure exerted by the earth's atmosphere. Standard sea-level pressure, by definition, equals 760 mmHg.

Table 2-1 Anatomical Dead Space in the Lungs Varies with Body Size

An estimate of anatomical dead space in milliliters is body weight expressed in pounds. If a subject weighs 180 lb, the dead space is close to 180 mL, or 0.18 L. Thus, for the average 500 mL or 0.5 L of air inspired per breath (V_T) at rest, 64% $[(0.5 \text{ L} - 0.18 \text{ L}) \div (0.5 \text{ L} \times 100)]$ ventilates the alveoli, and 36% remains in the dead space.

The respiratory zone extends from the terminal bronchioles to the alveolar ducts, alveolar sacs, and alveoli, where **gas exchange** between the lungs and blood takes place. First, the terminal bronchioles divide to form the respiratory bronchioles, which become even smaller bronchioles with some capacity for gas exchange as they ultimately give rise to alveolar ducts that end as clusters of thin-walled, inflatable alveolar sacs composed of alveoli. Macrophages in the alveoli protect the alveoli from foreign particles. Each alveolus is surrounded by a dense network of pulmonary capillaries that facilitate diffusion of oxygen and carbon dioxide.

Respiratory gas exchange by simple diffusion occurs between the alveoli and pulmonary capillaries. There are about 300 million to 500 million alveoli in the two lungs with an internal surface area for diffusion that is the equivalent of 60–80 m² (i.e., the size of a tennis court). Inspired oxygen diffuses from the alveoli to the pulmonary capillaries to gain entrance into the blood where it is bound to **hemoglobin** (Hb), forming **oxyhemoglobin** (HbO₂).

Mechanics of Ventilation

Ventilation is the movement of air into and out of the lungs. It requires a pressure gradient from outside of the body to the alveoli and, then, from the alveoli to the ambient air. The **barometric (atmospheric) pressure** is the outside pressure. At sea level, the pressure is 760 mmHg, or 760 torr. The pressure unit *torr* was named in honor of Evangelista Torricelli (1608–1647), who invented the barometer. If air is to move out of the lungs, the **intra-alveolar pressure** must exceed the barometric air pressure. The exchange of oxygen and carbon dioxide between the lungs and the atmosphere is referred to as **external respiration**.

The respiratory exchange of oxygen and carbon dioxide at the muscle tissue level is referred to as **internal respiration**. As the tissues consume oxygen, they produce carbon dioxide, which diffuses into the blood. This is termed **cellular respiration**. During rest, about 250 mL · min⁻¹ of oxygen leaves the alveoli to enter the blood, and about 200 mL · min⁻¹ of carbon dioxide diffuses from the blood into the alveoli. The $\dot{V}_{\text{CO}_2}/\dot{V}_{\text{O}_2}$ ratio, known as the **respiratory exchange ratio** (RER), increases from approximately 0.8 at rest (i.e., $200 \div 250 = 0.8$) to 0.9 or 1.0 or higher at moderate to maximum exercise. An RER of 0.8 means that approximately 80 molecules of CO₂ are exhaled from the lungs for every 100 molecules of O₂ that diffuse from the alveoli into the pulmonary capillary blood.

Inspiration

Inspiration begins with the contraction of the **diaphragm**, which forms the floor of the thoracic cavity. As the large, dome-shaped skeletal muscle contracts, it elongates the thoracic cavity and thereby increases the thoracic volume during inspiration. Three fourths of the work of enlarging the thoracic cavity during quiet inspiration is accomplished by the diaphragm. During more forceful inspirations, the muscles

between the ribs (**external intercostals**) along with other muscles lift the rib cage superiorly and laterally (Figure 2-3). Contraction of these muscles brings about a greater decrease in the pressure inside the **pleural cavity** (i.e., intrapleural pressure) to 754 mmHg as a result of expansion of the thorax. This causes the lungs to expand in size, which decreases the pressure inside the lungs (**intrapulmonary pressure**; also known as intra-alveolar pressure) below the atmospheric (barometric) pressure (757 mmHg vs. 760 mmHg). This pressure difference of 3 mmHg between the lungs and ambient air drives the air into the lungs. Other accessory muscles of inspiration include the sternocleidomastoid muscles, which originates from the manubrium sterni and medial third of the clavicle. These muscles are responsible for lifting the rib cage up and outward. During strenuous exercise, in particular, they contribute increasingly more to the inspiration along with the pectoralis major and minor muscles, the trapezius, and the serratus anterior muscles.

Expiration

Expiration begins with the relaxation of the diaphragm and external intercostals, which allows for a passive decrease in the thoracic volume. At the same time, the **internal intercostals** contract to cause the pressure inside the pleural cavity momentarily to increase, which decreases the alveolar volume and increases the intra-alveolar (intrapulmonary) pressure over the barometric air pressure. Air flows out

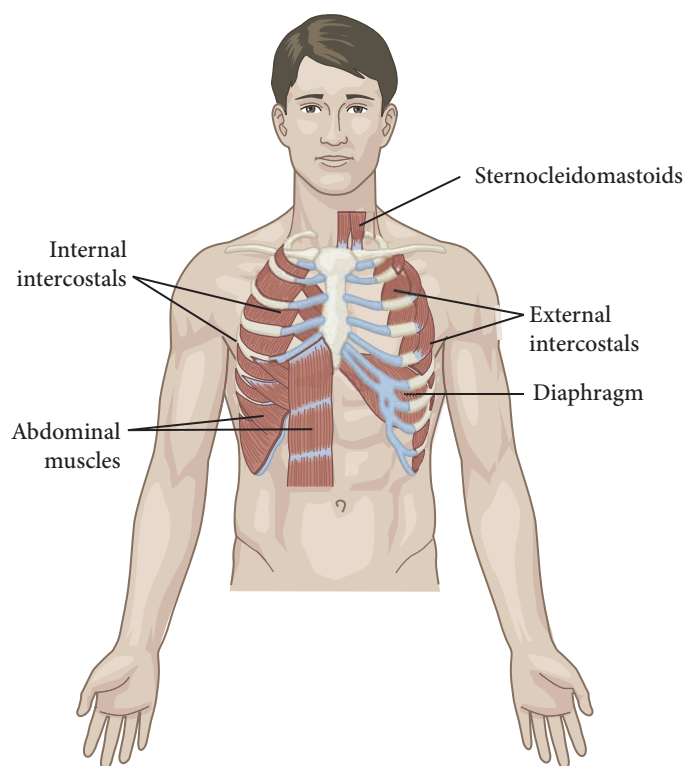


Figure 2-3 Muscles used for ventilation.

Intra-alveolar pressure

The pressure within the alveoli. When this pressure is less than the atmospheric pressure, air will move into the alveoli. When this pressure is greater than the atmospheric pressure, air will move out of the alveoli.

External respiration The exchange of respiratory gases between the air in the alveoli and the blood within the pulmonary capillaries.

Internal respiration The metabolic process by which cellular tissue absorbs oxygen from the arterial blood and releases carbon dioxide to the venous blood.

Cellular respiration The energy-releasing metabolic pathways in a cell that oxidize organic molecules such as glucose and fatty acids.

Respiratory exchange ratio The ratio between the amount of CO₂ exhaled and O₂ inhaled in one breath.

Diaphragm The primary respiratory muscle responsible for inspiration.

External intercostals Muscles originating between ribs 1 through 11 that are responsible for the elevation of the ribs, and expanding the transverse dimensions of the thoracic cavity to aid in quiet and forced breathing.

Pleural cavity The space between the serous membrane enveloping the lung and lining the internal surface of the thoracic cavity.

Intrapulmonary pressure The pressure within the alveoli that is sometimes called alveolar pressure.

Internal intercostals The internal intercostals counteract the action of the external intercostals. When they contract the ribs are brought closer together, which creates a positive pressure difference between the air in the lungs and the atmospheric air.

Intrapleural pressure An actual (or potential) space between the visceral pleural membrane covering the lungs and the somatic pleural membrane lining the thoracic wall.

of the lungs due to the pressure differential. During forced expiration such as during exercise when pulmonary ventilation is 15–20 times greater than at rest, contraction of the abdominal wall muscles (external oblique, internal oblique, transversus abdominis, and rectus abdominis) helps to increase intra-abdominal pressure to push the diaphragm upward. This decreases the vertical dimension of the thoracic cavity. The result is an increase in both the intrapleural and intra-alveolar pressures, which aids in expiration.

Pressure Changes

At the end of a normal respiration, **intrapleural pressure** (also known as the intrathoracic pressure) within the pleural sac is less than atmospheric (barometric) pressure. It averages 756 mmHg, or 4 mmHg less than barometric pressure (760 mmHg). The intrapulmonary pressure (or intra-alveolar pressure) is equal to barometric pressure. During inspiration, as described in **Box 2-1**, the volume of the thoracic cavity increases resulting in the expansion of the volume of gas in the pleural space. As the molecules move farther apart, the intrapleural pressure is decreased on average -2 mmHg (or to 754 mmHg). At the same time, the lungs are expanded, and gas molecules in the lung move farther apart. The intrapulmonary pressure is decreased below barometric air pressure (on average, -3 mmHg, or to 757 mmHg). This pressure difference of 3 mmHg between the atmospheric pressure and the intrapulmonary pressure causes the movement of air into the lungs. Alveolar pressure increases and becomes equal to barometric pressure at the end of inspiration.

During expiration, the steps are reversed. While the lungs are passively recoiling, the internal intercostals assist by contracting to decrease the size of the thoracic cavity further. The increase in intrapleural pressure now compresses the air molecules in the previously inflated lung into a smaller volume. Thus, the intrapulmonary (intra-alveolar) pressure is increased above the barometric (or atmospheric) pressure. As air flows out of the lungs down its pressure gradient, alveolar pressure decreases and becomes equal to barometric pressure at the end of expiration. Because a pressure

Box 2-1 Muscles Involved During Normal Inspiration and Forced Expiration

	Inspiration	Expiration
Normal breathing	The diaphragm is the primary muscle of respiration. Contraction and flattening of the diaphragm enlarges the chest cavity, thus lowering the air pressure within the lungs. Also, the contraction of the external intercostals helps to increase the thoracic and lung volume further, which decreases the intrapulmonary pressure.	Normal expiration is a passive process. When the diaphragm and the external intercostals relax along with the elastic recoil of the lungs, air inside the lungs is exhaled.
Forced breathing	In addition to what is required for normal inspiration, forced breathing requires additional nerve input via the phrenic nerves to the diaphragm and the nerves to the external intercostals and the sternocleidomastoid and scalene muscles to decrease the intrapulmonary pressure further to allow for a large inspired volume.	During forced expiration, the thorax and lungs return to their normal shape and size as the abdominal muscles and the internal intercostal muscles contract. The decrease in the size of the thoracic wall increases the intrapulmonary pressure, and air leaves the lungs.

gradient no longer exists, no movement of air can take place. The pressure in the pleural space remains negative to protect the lungs from collapsing. If the intrapleural pressure equals the atmospheric pressure, the lungs would collapse at the end of expiration.

Pulmonary Volumes and Capacities

There are primarily four factors that influence the measurement of **respiratory volumes and capacities**. They are physical conditioning, age, body size, and gender. As an example, because the lungs of a female are on average smaller than those of a male, the female's vital capacity is about 25% less. The same logic applies to a person who is 6 ft tall versus someone who is 5 ft tall. A well-trained male or female will have a larger vital capacity than an active or sedentary subject.

Lung Volumes and Capacities

For the most part, **lung volumes** are correlated with body size, age, and even height of the individual. A larger person tends to have a larger lung volume than that of a smaller person. An average resting V_T of a 70-kg (154-lb) subject is about 500 mL per breath ($\text{mL} \cdot \text{br}^{-1}$). That means the subject would breathe in 500 mL of air and breathe out 500 mL during each cycle of quiet breathing. Under no circumstances is the volume expired less than the inspired volume. The range in V_T is between 400 and 1000 $\text{mL} \cdot \text{br}^{-1}$ depending on the subject's size (Figure 2-4).

Respiratory volumes and capacities The volume of air associated with different phases of the respiratory cycle. Although lung volumes are directly measured, capacities are inferred from lung volumes.

Lung volumes The volume of air associated with different phases of the respiratory cycle.

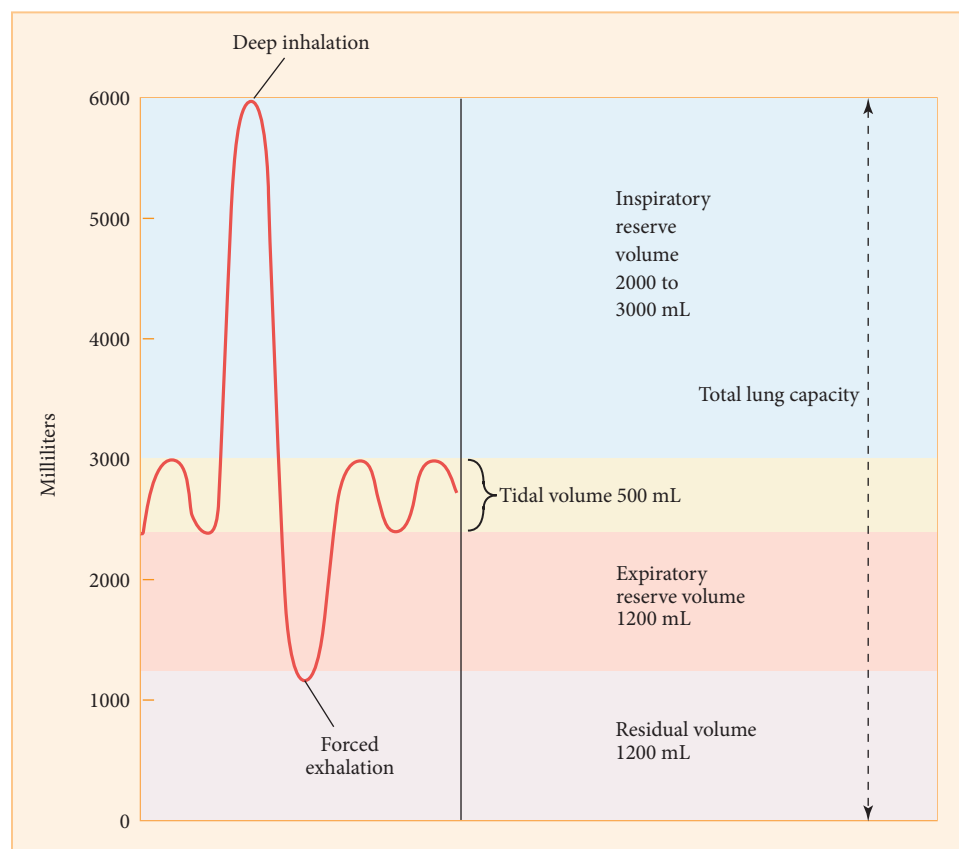


Figure 2-4 Respiratory volumes and capacities.

Inspiratory reserve

volume The maximal amount of additional air that can be drawn into the lungs after normal inspiration.

Expiratory reserve

volume The maximal volume of air that can be expelled from the lungs after normal expiration.

Vital capacity The greatest amount of air that can be exhaled following a maximal inhalation.

Forced vital capacity The amount of air that can be forcibly exhaled from the lungs after taking the deepest breath possible.

Residual volume The volume of air that remains in the lungs after a maximal expiratory effort.

Maximum voluntary

ventilation The greatest volume of gas that can be breathed per minute by voluntary effort.

At the end of a normal inspiration, an additional volume of air can be inhaled into the lungs. That volume is approximately 3000 mL during maximal inspiration. In a 70-kg man, this sixfold increase over the normal V_T is referred to as the **inspiratory reserve volume** (IRV). It requires contraction of the accessory respiratory muscles in addition to the diaphragm and external intercostal muscles. The reverse is true for the **expiratory reserve volume** (ERV). In this case, during the expiration phase of a normal V_T , the subject is asked to continue exhaling while forcing out of the lungs as much air as possible. On average, the range is between 1000 and 1200 mL for a 70-kg man. It will be lower in a woman. During exercise, V_T is increased because of the increased reliance on IRV and ERV, but primarily IRV.

When V_T is maximal, it is called **vital capacity** (VC). It represents the maximum volume of air exhaled after a maximal inspiration (i.e., the sum of the IRV, V_T , and ERV). As in V_T , vital capacity is correlated to body size. The average total air moved in one breath from a full inspiration is 4000 to 5000 mL in healthy men and a liter less in healthy women. The volume of air that remains in the lungs after a **forced vital capacity** (FVC) is the residual lung volume (RLV). It is approximately 1200 mL. Together, vital capacity and the **residual volume** (RV) constitute the total lung capacity ($TLC = VC + RV$). It is approximately 5800 mL.

To assess accurately the lungs of a person with severe lung disease, measurement of vital capacity by itself is not enough. Vital capacity can be normal, but the percentage of vital capacity achieved in 1 s ($FEV_{1.0}$) will be decreased. Another measure of pulmonary ventilation is $FEV_{1.0}$ to FVC, which reflects the expiratory power and resistance to expiration (i.e., the maximal airflow rate). With emphysema, asthma, chronic obstructive pulmonary disease (COPD), chronic bronchitis, or other bronchial restriction to air movement in the lungs, the $FEV_{1.0}/FVC$ ratio is decreased to one half that of an average response (i.e., 40 vs. 80% of FVC in 1 s).

The maximum amount of air that can be inhaled and exhaled in 1 min is termed **maximum voluntary ventilation** (MVV). It is of theoretical importance in assessing the maximal working capacity of the respiratory muscles. If the subject is a patient being tested, the MVV procedure is done over a 15-s time period before being extrapolated to a value for 1 min expressed as liters per minute. For healthy subjects, the range is 140–180 $L \cdot \text{min}^{-1}$ for men and 80–120 $L \cdot \text{min}^{-1}$ for women. Maximal pulmonary function can reach 200 $L \cdot \text{min}^{-1}$ in some athletes.

Pulmonary Ventilation

Pulmonary ventilation exchanges gases between the ambient air and the alveoli of the lungs. Ventilation, which is mechanical in nature, depends on a difference between the atmospheric air pressure and the pressure in the alveoli. Gases flow from areas of higher pressure to areas of lower pressure. For inspiration, the atmospheric pressure is greater than the alveolar pressure. The reverse is true for expiration.

Minute Ventilation

The volume of air inspired or expired per minute is known as \dot{V}_E . It is the product of an average breathing frequency (F_b) of 12 $\text{br} \cdot \text{min}^{-1}$ (also referred to as respiratory rate) and an average V_T of 500 $\text{mL} \cdot \text{br}^{-1}$. At rest, \dot{V}_E averages about 6000 $\text{mL} \cdot \text{min}^{-1}$ ($12 \text{ br} \cdot \text{min}^{-1} \times 500 \text{ mL} \cdot \text{br}^{-1}$) or 6 $L \cdot \text{min}^{-1}$ ($6000 \text{ mL} \cdot \text{min}^{-1} \div 1000 \text{ mL} \cdot L^{-1} = 6.0 L \cdot \text{min}^{-1}$) with a range between 5 and 8 $L \cdot \text{min}^{-1}$. With exercise, \dot{V}_E is increased as a direct function of the oxygen needed at the cell level and the carbon dioxide produced by the muscles. Either an increase in F_b or V_T or both will increase \dot{V}_E .

Of course, it is better to increase \dot{V}_E during exercise with an increase in V_T rather than F_b . For example, the \dot{V}_E for subject A at 600 kpm on the bicycle ergometer is $46 \text{ L} \cdot \text{min}^{-1}$, which is the product of a V_T of 2 L per breath and an F_b of $23 \text{ br} \cdot \text{min}^{-1}$. Subject B at the same workload has the same \dot{V}_E , but his product values are very different. His V_T is 1.6 L per breath, and his F_b is $29 \text{ br} \cdot \text{min}^{-1}$. Both subjects are processing oxygen to produce energy for muscle contraction. The difference is that the respiratory effort for subject B is much greater in that his \dot{V}_E is dependent upon F_b given the reduced V_T . This alone increases the energy requirement of the respiratory muscles, which ultimately detracts from the oxygen available to the muscular system.

During maximal exercise, the F_b of healthy subjects is about $40 \text{ br} \cdot \text{min}^{-1}$. Their V_T is about 2.6 L per breath with a \dot{V}_E of $104 \text{ L} \cdot \text{min}^{-1}$ to help sustain a $\dot{V}_{O_2 \text{ max}}$ of about $3 \text{ L} \cdot \text{min}^{-1}$. In contrast, highly trained athletes can achieve F_b values of about 60 breaths $\cdot \text{min}^{-1}$ and a V_T of 3 L per breath to produce a substantial increase in \dot{V}_E of $180 \text{ L} \cdot \text{min}^{-1}$. In terms of the **maximal minute ventilation** ($\dot{V}_{E \text{ max}}$) that is critical to achieving a high $\dot{V}_{O_2 \text{ max}}$ value, the $\dot{V}_{E \text{ max}}$ is ~ 15 times the resting value of $7 \text{ L} \cdot \text{min}^{-1}$. In the highly trained athlete, $\dot{V}_{E \text{ max}}$ can reach $180 \text{ L} \cdot \text{min}^{-1}$, which is a 26-fold increase over this athlete's average resting \dot{V}_E (Box 2-2).

Alveolar Ventilation

Given the anatomical design of the lungs, at first it was expected that all of the air that enters the lungs reaches the alveoli. Yet, it is clear this is not the case. A small portion of the V_T remains in the dead space (DS). This means that not all of the \dot{V}_E effort takes part in the exchange of oxygen for carbon dioxide at the **alveolar-capillary membrane**. That part of the V_T that does not reach the alveoli remains in the conducting airways collectively termed the anatomical dead space (nose, mouth, trachea, bronchi, and bronchioles).

For a subject who weighs 150 lb, the volume of air that does not participate in gas exchange is about 0.25 of the resting \dot{V}_E of $6 \text{ L} \cdot \text{min}^{-1}$ or $6000 \text{ mL} \cdot \text{min}^{-1}$, which equals $1500 \text{ mL} \cdot \text{min}^{-1}$ (6000×0.25), or an average of $150 \text{ mL} \cdot \text{br}^{-1}$ ($1500 \div 10 \text{ br} \cdot \text{min}^{-1}$). At rest the F_b is 10 to $14 \text{ br} \cdot \text{min}^{-1}$. Note that this volume in the anatomical dead space approximates body weight in pounds. For a subject weighing 250 lb, the estimate for anatomical dead space in the calculation of alveolar ventilation (\dot{V}_A) is 250 mL. In this case, the $250 \text{ mL} \cdot \text{br}^{-1}$ is not used for gas exchange.

Alveolar ventilation is the most important variable in gas exchange. It is calculated by subtracting the anatomical dead space from V_T , which is then multiplied by

Box 2-2 Mean Values in Healthy Young Men for Ventilation and Oxygen Consumption

Parameter	Rest	Low	Moderate	Maximum (Untrained)	Maximum (Athlete)
\dot{V}_{O_2} ($\text{L} \cdot \text{min}^{-1}$)	0.25	1.6	2.6	3.4	~ 5
\dot{V}_{CO_2} ($\text{L} \cdot \text{min}^{-1}$)	0.20	0.85	2.4	3.5	6
\dot{V}_E ($\text{L} \cdot \text{min}^{-1}$)	6	30.6	71.3	116.1	180
V_T (L)	0.5	1.7	2.3	2.7	3.0
F_b (breaths $\cdot \text{min}^{-1}$)	12	18	31	43	60

Maximal minute ventilation

The maximum amount of air inspired or expired each minute. It is calculated as the product of tidal volume and frequency of breaths.

Alveolar-capillary membrane

The alveolar-capillary membrane (or barrier) exists in the gas exchange region of the lungs to prevent air bubbles from forming in the blood and blood from entering the alveoli. The thin tissue barrier is permeable to oxygen in the alveolar air and carbon dioxide from the blood in the pulmonary capillaries.

Alveolar ventilation The volume of air available for gas exchange; it is calculated as tidal volume minus dead space times frequency.

the Fb. For example, for a 150-lb subject at rest, that part of \dot{V}_E that mixes with air in the alveoli is $4200 \text{ mL} \cdot \text{min}^{-1}$ ($500 \text{ mL} \cdot \text{br}^{-1} - 150 \text{ mL} = 350 \text{ mL} \cdot \text{br}^{-1} \times 12 \text{ br} \cdot \text{min}^{-1} = 4200 \text{ mL} \cdot \text{min}^{-1}$). This means that 70% of the \dot{V}_E mixes with alveolar air. What is important to recognize is that \dot{V}_E by itself does not mean that an adequate gaseous exchange is maintained. To better understand this point, what would happen if the size of the dead space is increased? A dead-space air of 200 mL per breath would decrease resting \dot{V}_A to $3600 \text{ mL} \cdot \text{min}^{-1}$. This means that 60% of the \dot{V}_E mixes with alveolar air. The dead-space air has increased to 40% of \dot{V}_E . Thus, \dot{V}_E remained the same, but \dot{V}_A decreased. This result can have a negative effect on gas exchange.

Conversely, if the subject's resting V_T increased from 500 to 600 $\text{mL} \cdot \text{br}^{-1}$ while \dot{V}_E remained unchanged, \dot{V}_A is increased ($6000 \div 600 = 10 \text{ breaths} \cdot \text{min}^{-1}$). Alveolar ventilation is now $4500 \text{ mL} \cdot \text{min}^{-1}$ or an increase of 20% in the \dot{V}_E that mixes with the alveolar air. Stated somewhat differently, this means that 80% of the inspired air per minute ventilates the alveoli. The lungs are more efficient when V_T is increased than when relying on an increase in breathing rate.

Similarly, with regard to \dot{V}_E during exercise, the size of the V_T is more important than Fb. Because anatomical dead space is constant, the increased V_T at a given breathing frequency is the most efficient means of increasing the volume of air exchanged between the atmosphere and alveoli per minute. Although this is true, there is the assumption that all the atmospheric air entering the alveolar sacs participates in gas exchange with the blood in the capillaries. What if the alveoli are not equally ventilated with air or only partly in contact with capillary blood? If one or both occurred, the assumption of a match between air and blood would be wrong.

The point is that it is always likely that some unknown percentage of ventilated alveoli is not adequately perfused with blood. This is termed **alveolar dead space** (or physiologic dead space). The important thing to remember is that among healthy subjects, alveolar dead space is of little importance.

Alveolar dead space That part of the physiological dead space that participates in gas exchange, except for the underperfused or nonperfused alveoli.

Gas Exchange in the Lungs and the Body

Gas Exchange

Gas exchange takes place primarily in two areas of the body. The first exchange is between the alveoli and the pulmonary capillaries. Oxygen from the lungs diffuses into the blood, and carbon dioxide in the blood diffuses into the lungs. The force behind the diffusion is the partial pressure of each gas. Because oxygen moves from the lungs into the blood, the pressure exerted by the oxygen gas molecules causes the oxygen to enter the blood where the concentration of oxygen and the pressure are smaller.

The opposite takes place with carbon dioxide. At the cell level, given the higher partial pressure of oxygen in the blood versus the tissues, oxygen dissociates from the hemoglobin and moves from the blood into the **mitochondria** of the tissues. The high metabolic state of the muscles increases the concentration of carbon dioxide within the tissues. This means the partial pressure of carbon dioxide (P_{CO_2}) is higher than the arterial blood carbon dioxide (P_{aCO_2}) that surrounds the muscle cells, so carbon dioxide leaves the cells and enters the venous side of the arterial system to be transported to the lungs.

Mitochondria Cellular organelles that contain the enzymes for the Krebs cycle, electron transport chain, and the fatty acid cycle that are involved in producing ATP.

Ambient Air

Understanding gas exchange begins with acknowledging the concentration of gases in the air and their individual pressures. **Ambient (atmospheric) air** is a mixture of oxygen, carbon dioxide, nitrogen, water vapor, and small quantities of inert gases. Oxygen makes up 20.93% of the ambient air, carbon dioxide is 0.03%, and nitrogen is 79.04% of the ambient air. Each gas consists of molecules that collide with each other. The pressure exerted by a gas is dependent on the number of collisions. The higher a gas concentration is in a given volume, the higher the partial pressure of the gas. Because oxygen is not the only gas in ambient air, its pressure is only part of the total pressure of the mix of gases.

The partial pressure of oxygen, like other gases, is dependent on its fractional concentration and the barometric pressure. At sea level, oxygen equals 20.93% of the total 760 mmHg pressure exerted by the air mixture. This means oxygen exerts a partial pressure of 159 mmHg ($760 \text{ mmHg} \times 0.2093$). Carbon dioxide exerts a pressure of 0.3 mmHg, and nitrogen exerts the largest pressure ($760 \text{ mmHg} \times 0.7904 = 600.7 \text{ mmHg}$). It is common practice to designate the partial pressure of individual gases in a mixture with a capital “P” followed by the symbol for the gas. Thus, the partial pressure of oxygen is P_{O_2} , carbon dioxide is P_{CO_2} , and nitrogen is P_{N_2} .

In the Trachea

As the ambient air enters the respiratory tract, it becomes saturated with water vapor. This process dilutes the inspired air mixture. At body temperature, the partial pressure of water (P_{H_2O}) molecules in humidified air equals 47 mmHg. This value must be subtracted from 760 mmHg ($760 - 47 = 713 \text{ mmHg}$). This means the partial pressure of the inspired oxygen at 159 mmHg is decreased in the trachea ($713 \text{ mmHg} \times 0.2093$) to 149 mmHg. At this point, this is not a problem because the partial pressure of oxygen in the pulmonary capillaries is about 40 mmHg at rest. There is still a considerable pressure differential to ensure that the oxygen leaves the alveoli to enter the blood.

Alveolar Air

The composition of alveolar air is fairly constant. It is saturated with water vapor and contains a higher concentration of carbon dioxide than that of the atmospheric air. Gas exchange between the blood and the alveoli is continuous. The alveoli always have gases in them, thus making the alveoli independent of the respiratory activity. The driving pressure for oxygen and carbon dioxide is the partial pressure of each gas, where partial pressure is the product of the percentage gas concentration and total pressure of the gas mixture. As it turns out, the percentage of oxygen concentration in the alveoli averages about 14.5%. This means the average alveolar partial pressure of oxygen (P_{AO_2}) is 103 mmHg ($760 - 47 = 713 \text{ mmHg} \times 0.145$), or an alveolar average of about 100 mmHg (Box 2-3).

Even though the partial pressure of oxygen is about 60 mmHg lower than that of the dry ambient air, the partial pressure of 100 mmHg produces an average driving pressure of 60 mmHg for oxygen to diffuse from the alveoli into the venous blood (i.e., an area of lower partial pressure and an average P_{O_2} of 40 mmHg). Because the pulmonary blood has a higher partial pressure of carbon dioxide ($P_{CO_2} = 46 \text{ mmHg}$ on average) than that of the alveoli, carbon dioxide molecules leave the blood to enter the alveoli. This increases the percentage of carbon dioxide in the alveoli from 0.03%

Ambient (atmospheric) air

Ambient air is the outdoor air in which humans and other organisms live and breathe.

Box 2-3 Gas Partial Pressures in Ambient Air (Sea Level) and Alveolar Air (Sea Level)

Gases	Percentage (%)	P _B (mmHg)	P (mmHg)
Ambient air			
O ₂	20.93	760	159*
CO ₂	0.03	760	0.3
N ₂	79.04	760	600.7
H ₂ O	0.00	760	0.0
Total			760
Alveolar air			
O ₂	~13.7	760 – 47 = 713 ⁺	~104
CO ₂	~5.3	760 – 47 = 713	~40
N ₂	~79.8	760 – 47 = 713	~569
H ₂ O			47
Total			760

Note: The partial pressure (P) of a gas is a measure of the concentration of a gas in a mixture of gases. Atmospheric pressure at sea level is approximately 760 mmHg. P_B stands for barometric pressure.

*Oxygen constitutes 21% of the atmosphere, so its partial pressure is 21% of 760, or 159 mmHg.

⁺The P_{H₂O} of 47 mmHg is subtracted from 760 mmHg before calculating P_{O₂}, P_{CO₂}, and P_{N₂}.

in the ambient air entering the alveoli to an average of 5.5% in the alveolar air (713×0.055). Thus, the average partial pressure of carbon dioxide (P_{CO₂}) is equal to 39 mmHg. In general, the alveolar partial pressure of carbon dioxide (P_{ACO₂}) is taken to be 40 mmHg. This represents a small pressure gradient of 6 mmHg for carbon dioxide to diffuse from the blood (P_{CO₂} = 46 mmHg) to the lungs.

In Tissues

As the blood leaves the lungs to the left side of the heart to be pumped to the tissues, blood arriving at the tissues has been normalized for its carbon dioxide and saturated with oxygen. The partial pressures for oxygen and carbon dioxide are approximately 100 mmHg and 40 mmHg, respectively. **Cellular metabolism** is dependent upon both gases in the arterial blood remaining in their respective ranges. At rest, the average P_{O₂} within the muscles is around 40 mmHg while the P_{CO₂} is about 46 mmHg. This means that there is a driving force of 60 mmHg for oxygen to leave the blood through the capillary wall and enter the tissues. Likewise, because the P_{CO₂} is higher in muscle than in the arterial blood, carbon dioxide diffuses from the muscles to the blood. This increases the P_{CO₂} of the venous blood to an average of 46 mmHg. The diffusion of oxygen from the arterial blood to the muscle cell leaves the venous blood with a P_{O₂} of 40 mmHg.

As the venous blood from the tissues leaves the right ventricle on its way to the lungs, the P_{CO₂} of 46 mmHg at rest relative to the alveolar P_{CO₂} of 40 mmHg drives the carbon dioxide from the pulmonary capillaries into the alveoli. During exercise, carbon dioxide pressure in the tissues increases to 70, 80, or even 90 mmHg depending upon the intensity of the exercise. If the venous P_{CO₂} is 70 mmHg, the driving

Cellular metabolism The chemical processes by which cells produce the substances and energy to sustain life.

pressure to move carbon dioxide into the lungs to be exhaled would be 30 mmHg (i.e., 70 mmHg, venous – 40 mmHg, alveolar). This is a fivefold increase in pressure, thus helping to ensure that with exercise and the faster movement of blood through the **pulmonary capillaries**, the excess carbon dioxide is moved to the lungs in a timely fashion. The opposite is true with oxygen, given the P_{O_2} of 100 mmHg in the alveoli and possibly 20 or even 10 mmHg in the venous blood (which reached equilibrium with the pressure in the active muscle tissue). The pressure differential of 80 or 90 mmHg between the alveoli and the capillary blood results in a very fast diffusion of oxygen into the blood.

Pulmonary capillaries

Tiny blood vessels that surround each alveolus for the purpose of facilitating the alveoli exchange of oxygen for carbon dioxide.

Oxygen and Carbon Dioxide Transport

The transport of oxygen and carbon dioxide occurs both via the liquid portion of the blood (plasma) and the hemoglobin (Hb) in red blood cells (erythrocytes). Only a very small amount of oxygen is transported in the plasma; this is dissolved oxygen. It would be impossible to live on this volume. This is not the case with oxygen that diffuses across the respiratory membrane into blood to combine with hemoglobin to form oxyhemoglobin (HbO_2). This method of transport of oxygen increases the oxygen carrying capacity 65 times above that of plasma.

Stated somewhat differently, approximately 98.5% of the oxygen transported in the blood is in combination with hemoglobin in red blood cells, and the remaining 1.5% is dissolved in the water part of the plasma. As the tissue cells use oxygen in aerobic respiration, they produce carbon dioxide that diffuses from the cells to the blood. At the venous end of the capillaries, carbon dioxide is transported dissolved in the plasma, in combination with hemoglobin, and in the form of bicarbonate ions.

Oxygen Transport in the Blood

In Physical Solution

About 0.3 mL of oxygen is dissolved in the plasma of 100 mL of blood (or $3 \text{ mL} \cdot \text{L}^{-1}$). As the average volume of blood is about 5 L, a very small amount of oxygen is dissolved in the fluid portion of the blood ($3 \text{ mL} \cdot \text{L}^{-1} \times 5 \text{ L} = 15 \text{ mL}$). Because oxygen consumption ($\dot{V}O_2$) at rest is about $250 \text{ mL} \cdot \text{min}^{-1}$, the dissolved oxygen of 15 mL is not enough to sustain life. In fact, cardiac output (\dot{Q}) would have to increase 16.7 times the resting blood flow per minute (e.g., $\dot{Q}, 5 \text{ L} \cdot \text{min}^{-1} = 250 \text{ mL} \cdot \text{min}^{-1}$, thus $250 \div 15 = 16.7$, then, $5 \times 16.7 = 83.5 \text{ L} \cdot \text{min}^{-1}$). Obviously, the heart cannot pump $84 \text{ L} \cdot \text{min}^{-1}$. What then is the purpose of the oxygen transported in the physical solution? The dissolved oxygen establishes the P_{O_2} of the arterial and venous blood (i.e., the amount dissolved is directly proportional to the P_{O_2} of the blood). In other words, as mentioned earlier, at an arterial P_{O_2} of 100 mmHg, only 3 mL of oxygen can dissolve in 1 L of blood. Thus, the oxygen bound to hemoglobin does not contribute to the P_{O_2} of the blood.

Combined with Hemoglobin

Hemoglobin (Hb) in red blood cells contains iron (heme) and protein (globin). Each of the four iron atoms within the heme portions of a hemoglobin molecule can combine with one oxygen molecule to form oxyhemoglobin. In men, each 100 mL of blood contains an average of 15 g of Hb. Women have slightly less hemoglobin

per 100 mL of blood, about 13 g. Because oxygen is transported in the blood to the tissues by hemoglobin, the higher oxygen transport capacity of men gives them a physiologic advantage over women. As an example, when 1 g of Hb combines with oxygen (i.e., becomes saturated with O_2), it is the equivalent of 1.34 mL of oxygen transported by the blood. Thus, it is a matter of multiplying hemoglobin (in grams per 100 mL of blood) by the oxygen capacity of hemoglobin. In the case of men, 15 g of Hb per 100 mL of blood ($15 \text{ g} \cdot 100 \text{ mL}^{-1}$) equals 20 mL of oxygen ($15 \times 1.34 = 20.1$) per each 100 mL of blood (or 20 volumes percent). In contrast, women have the same capacity to saturate the hemoglobin with oxygen but have less Hb per 100 mL of blood (e.g., 13 g per 100 mL $\times 1.34 \text{ mL} = 17.4 \text{ mL}$ of oxygen for every 100 mL of blood).

To achieve full oxygen saturation of hemoglobin ($SbO_2\%$) means that each of the four heme groups is bound to oxygen. Naturally, it is important fully to saturate the hemoglobin that is exposed to alveolar gas. To determine the percentage saturation of hemoglobin with oxygen ($SbO_2\%$), the total oxygen actually combined with hemoglobin is divided by the oxygen carrying capacity of hemoglobin. For example, if oxygen capacity is 20 volumes percent and the actual amount of oxygen combined with hemoglobin is 15 volumes percent, then $SbO_2\%$ is $15 \div 20 \times 100$, which equals 75%. This means that the arterial blood is transporting 25% less oxygen than it should transport for each 100 mL of blood pumped from the left ventricle. Hemoglobin that is not bound to oxygen is referred to as **deoxyhemoglobin** (or reduced Hb).

Because it is important to saturate hemoglobin with oxygen fully, what are the factors that affect achievement of oxygen saturation? To begin with, it should be clear by now that the PO_2 of blood is especially important. Then, too, the temperature of the blood, the pH (acidity) of the blood, and the amount of carbon dioxide in the blood play a role in the percentage saturation of hemoglobin. To understand these relationships better, it is helpful to consider them singly and collectively using the oxyhemoglobin dissociation curve.

Oxyhemoglobin (HbO_2) Dissociation Curve

The **oxyhemoglobin dissociation curve** is an S-shaped curve that represents the percent saturation of hemoglobin in relation to oxygen pressure. The right y -axis shows the quantity of oxygen carried in each 100 mL of blood under normal resting conditions (i.e., with blood pH of 7.4 and with a temperature of 37°C). The x -axis shows the PO_2 (mmHg) in solution. Note that the arterial blood PO_2 is 100 mmHg. This occurs at the alveoli–capillary level during the time the blood moves through the capillaries. At a PO_2 of 100 mmHg, the 15 g of Hb per 100 mL of blood is fully saturated with oxygen. That means the arterial blood is transporting 20 mL of oxygen for each 100 mL of blood to the tissues. At the cell level, the PO_2 is 40 mmHg at rest. The pressure difference of 60 mmHg causes the oxygen to dissociate from hemoglobin to move into the cell. This means that 5 mL of oxygen (or $5 \text{ mL } O_2 \cdot \text{dL}^{-1}$) went into the muscle, or 25% of the arterial oxygen per 100 mL of blood was consumed for metabolic purposes. The remaining difference is 15 mL of oxygen per 100 mL of blood (or $15 \text{ mL } O_2 \cdot \text{dL}^{-1}$), which corresponds to 75% saturation of hemoglobin in the venous blood. The 5 mL of oxygen per 100 mL of blood is referred to as the **arteriovenous oxygen difference** ($a-vO_2$ diff). That is, $a-vO_2$ diff is the difference between the amount of oxygen carried in the arterial blood and the amount of oxygen returned in the venous blood. Thus, it represents the amount of oxygen released (dissociated) from the red blood cells (also called the coefficient of oxygen utilization) that is used by the tissues to support cellular metabolism (Figure 2-5).

Deoxyhemoglobin The hemoglobin in the blood that is not in combination with oxygen.

Oxyhemoglobin dissociation curve A graph that identifies the amount of oxygen chemically bound to the hemoglobin in blood as a function of oxygen pressure.

Arteriovenous oxygen difference The difference between oxygen content in the arterial blood and the amount of oxygen in the venous blood.

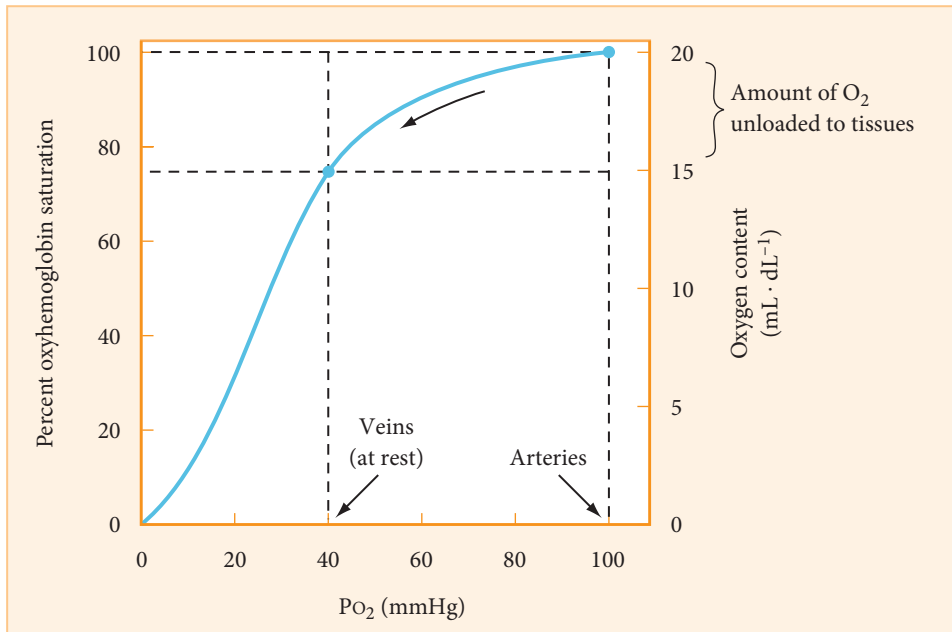


Figure 2-5 Oxyhemoglobin saturation and blood oxygen content.

Whereas the large change in PO₂ shown by the shallow, upper part of the curve is associated with a small change in the amount of oxygen held by hemoglobin, the smaller change in PO₂ shown by the steep middle and lower portions of the curve is associated with a large change in the amount of oxygen held by hemoglobin, reflecting a greater cellular need for oxygen. From about a PO₂ of 50 mmHg, a small change in PO₂ is associated with a large change in **hemoglobin saturation**. For example, if tissue PO₂ decreases from 40 to 20 mmHg, S_bO₂% decreases from 75 to 20%. This is an additional a-vO₂ diff of 11 mL of oxygen consumed at the cell level in addition to the original 5 mL of oxygen, or a total of 16 mL of oxygen per 100 mL of blood. The challenge in consumption of oxygen exists at the cell level. Oxygen cannot be consumed if the mitochondria are not using the oxygen in the first place. If the mitochondria can use the oxygen, given the level of intensity of muscle contraction, then the PO₂ within the muscle cell can decrease. This cellular adjustment allows for an increased dissociation of oxygen from hemoglobin. More oxygen means more energy for sustaining or increasing aerobic performance.

As mentioned earlier, the oxyhemoglobin dissociation curve is also under the influence of increases in blood acidity (which decreases pH), temperature, and carbon dioxide. Each of the three can cause a shift of the curve to the right, thus decreasing the tendency for oxygen to remain bound to hemoglobin. This is known as the **Bohr effect** after its discoverer, Christian Bohr. The influence of these three factors is particularly important during exercise, in which there is an increase in carbon dioxide production, an increase in lactic acid, which decreases the blood pH, and an increase in body temperature due to elevated temperatures resulting from increased metabolism. Hence, at a given PO₂, a shift of the oxygen–hemoglobin curve to the right makes more oxygen available to the tissues for use in cellular metabolism. The increase in oxygen could be as high as 75–85%, which would represent an arterio-venous oxygen difference of 15–17 mL of oxygen per 100 mL of blood (or 15–17 mL O₂ · dL⁻¹).

Hemoglobin saturation

The extent to which hemoglobin is loaded with oxygen molecules.

Bohr effect The effect of blood pH on the dissociation of oxyhemoglobin whereby dissociation is promoted by a decrease in the pH.

2,3-Diphosphoglycerate

Also known as 2,3-Bisphosphoglycerate or 2,3-BPG.

2,3-Bisphosphoglycerate

A product of red blood cells that increases the ability of oxyhemoglobin to dissociate and release its oxygen.

Another factor that influences hemoglobin's affinity for oxygen is the anaerobic metabolite **2,3-diphosphoglycerate** (2,3-DPG; formerly called diphosphoglycerate). This erythrocyte constituent is also referred to as **2,3-bisphosphoglycerate** (BPG). It is produced in the red blood cell as a by-product of glycolysis because the red blood cell lacks a nucleus and mitochondria, thus it cannot use oxygen. An increase in 2,3-DPG shifts the oxygen hemoglobin dissociation curve to the right: It facilitates oxygen dissociation by combining with subunits of hemoglobin to decrease the affinity of hemoglobin for oxygen (just as it increases in CO_2 and decreases in pH). The end result is that more oxygen is unloaded as the blood flows through the tissues. With more oxygen at the cell level, there is an increase in aerobic performance.

Today, it is common practice for endurance athletes to use a combination of high-altitude living and low-altitude training (known as Hi-Lo training) to increase RBC mass and the RBC's content of 2,3-DPG. The 2,3-DPG binds with the Hb molecule to reduce the affinity of Hb for oxygen while also decreasing RBC pH. Both conditions cause a rightward shift in the dissociation, which promotes oxygen unloading at the active muscles and theoretically increases the oxygen utilization curve. The exact mechanisms of improvement in athletic performance are yet to be fully understood, especially as some athletes do not respond to the Hi-Lo training.

Myoglobin and Storage of Oxygen

Myoglobin A protein in muscle fibers that binds oxygen and releases it at low PO_2 values.

Like hemoglobin, the molecule **myoglobin** (Mb) is an iron-protein compound that combines with oxygen. It is found in cardiac muscle cells and skeletal muscle fibers, especially in the slow-twitch fibers (i.e., high aerobic capacity). Its purpose is to help with the delivery of oxygen to the mitochondria once the oxygen is released from hemoglobin at the tissues. Because myoglobin has a higher affinity for oxygen than that of hemoglobin, it encourages the unloading of oxygen from hemoglobin. The binding of oxygen to myoglobin (MbO_2) acts as a temporary storage of oxygen in the muscle cell, especially under PO_2 conditions in which there is already an adequate supply of oxygen to the cell. Otherwise, when the PO_2 of the cell reaches very low values (e.g., 10 or 5 mmHg) during vigorous exercise, myoglobin unloads its oxygen to the mitochondria within the muscle fiber. Also, an increase in myoglobin increases mitochondrial respiration, particularly during intermittent periods of low PO_2 in the arterial blood.

Carbon Dioxide Transport in the Blood

The arterial blood transports approximately $49 \text{ mL CO}_2 \cdot \text{dL}^{-1}$ (or 49 mL per 100 mL of blood), whereas the venous blood transports approximately $54 \text{ mL CO}_2 \cdot \text{dL}^{-1}$. As the end product of oxidative metabolism, carbon dioxide is transported in three ways in the blood: (1) dissolved in physical solution; (2) as bicarbonate; and (3) as carbamino compounds (**Figure 2-6**). These three forms of CO_2 are in equilibrium with one another, and it is the dissolved fraction in physical solution (plasma) that exerts the partial pressure measured as PaCO_2 . Normal PaCO_2 ranges between 36 and 44 mmHg, whereas the mixed venous partial pressure of carbon dioxide ($\text{P}\bar{\text{v}}\text{CO}_2$) is approximately 6 mmHg higher. Unlike oxygen, where the partial pressure at rest is high in the arterial blood (PO_2 of 100 mmHg) and low in the venous blood (PO_2 of 40 mmHg), the partial pressure of carbon dioxide is higher in the venous blood (PCO_2 of 46 mmHg) than in the arterial blood (PCO_2 of 40 mmHg).

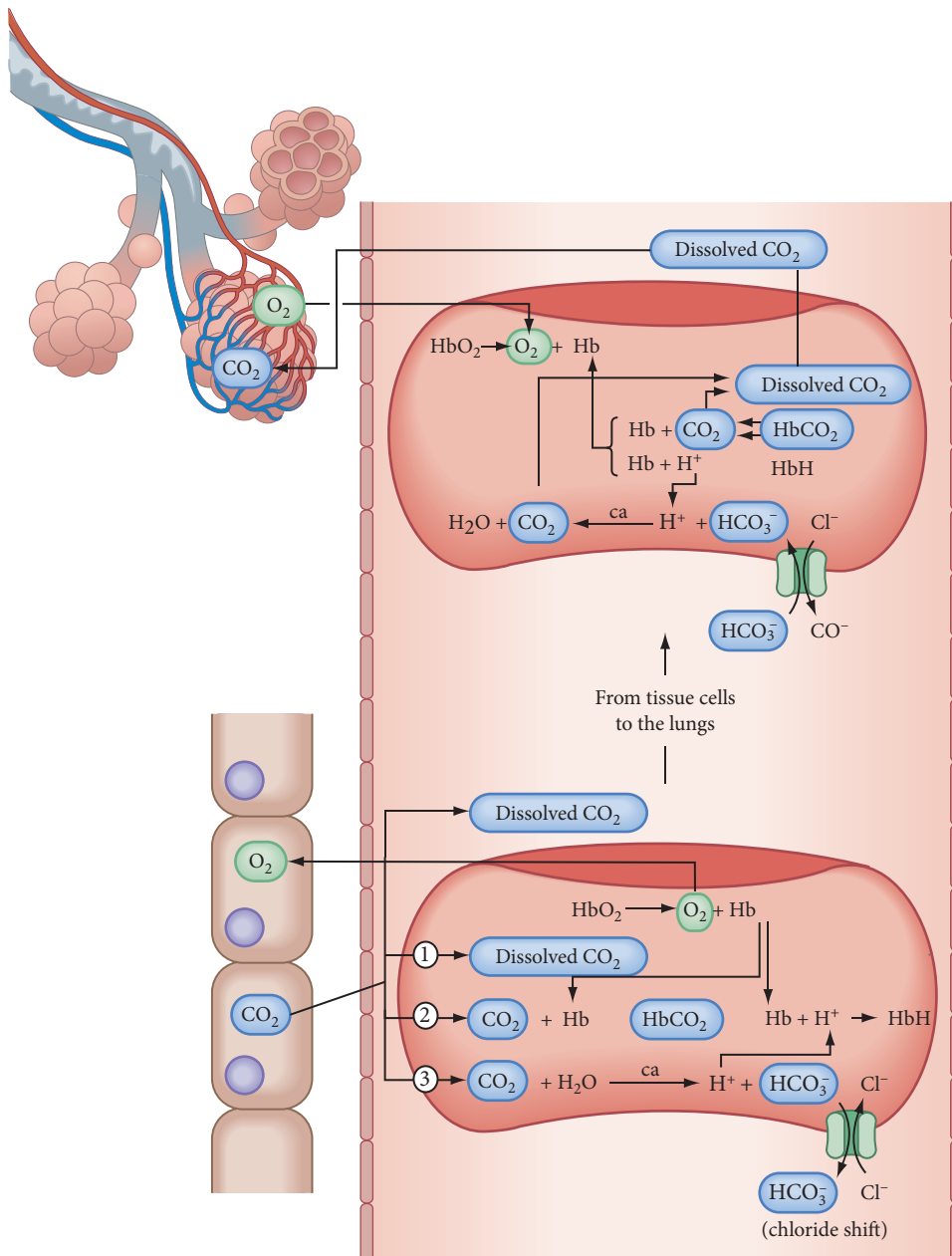


Figure 2-6 Carbon dioxide transport in the blood (Ca, carbon anhydrase).

In Solution

The amount of dissolved CO_2 in the venous blood is small but larger than that of oxygen because it is more soluble than oxygen in plasma. Thus, a greater proportion of the total CO_2 than of O_2 is dissolved in the blood. **Dissolved CO_2** represents a range of 10% of the total carbon dioxide transported from the cells to the lungs.

Dissolved CO_2 The simplest method of carbon dioxide transport is to physically dissolve it in blood plasma where it diffuses into the lungs and is exhaled from the body.

Carbonic acid A weak acid that is formed when carbon dioxide (CO_2) is dissolved in water.

Carbaminohemoglobin

When carbon dioxide binds to hemoglobin, carbaminohemoglobin is formed, lowering hemoglobin's affinity for oxygen via the Bohr effect.

Haldane effect While the increased oxygenation of hemoglobin promotes dissociation of carbon dioxide, deoxygenated hemoglobin has a greater affinity for CO_2 than it does for oxygen.

Carbonic anhydrase An enzyme that catalyzes the formation or breakdown of carbonic acid. When carbon dioxide concentrations are high, it catalyzes the formation of carbonic acid from CO_2 and H_2O . When carbon dioxide concentrations are low, the breakdown of carbonic acid to CO_2 and H_2O is catalyzed.

Bicarbonate ion The main form by which carbon dioxide is transported in the blood. It is an important extracellular buffer, which helps neutralize the effects of hydrogen ions produced during anaerobic respiration.

Potassium chloride The chemical compound potassium chloride (KCl) is a metal halide salt composed of potassium and chlorine.

Chloride shift The movement of chloride ions from the plasma into the red blood cells as a result of the transfer of carbon dioxide from the tissues to the plasma; a process that helps to maintain blood pH.

In the blood, carbon dioxide reacts with water to form a weak acid, **carbonic acid** (H_2CO_2). What is important about the dissolved CO_2 is that the amount of CO_2 physically dissolved in the plasma depends on the P_{CO_2} .

As Carbamino Compounds

Given the higher carbon dioxide concentration in the tissues resulting from energy metabolism versus the P_{CO_2} in the blood flow to the muscles, carbon dioxide diffuses into the venous blood. Once carbon dioxide enters the blood, it is carried in chemical combination with the globin of hemoglobin to form a carbamino compound while releasing a proton: $\text{Hb} \cdot \text{NH}_2 + \text{CO}_2 \rightleftharpoons \text{Hb} \cdot \text{NH} \cdot \text{COOH}$. The carbamino compound formed is called **carbaminohemoglobin** (HbCO_2). This occurs very rapidly without an enzyme. Reduced Hb combines more CO_2 than HbO_2 . It represents approximately 30% of carbon dioxide that is transported to the lungs.

At the lungs, the formation of carbaminohemoglobin is reversed, thus allowing carbon dioxide to diffuse from the pulmonary capillary blood into the alveoli from which the carbon dioxide is exhaled. Simultaneously, as oxygen diffuses from the alveoli into the pulmonary capillaries (thus, oxygenating Hb), it decreases hemoglobin's capacity to bind carbon dioxide. At the tissues, after hemoglobin has released oxygen, the hemoglobin has an increased ability to pick up carbon dioxide. This cycle of events facilitates the removal of carbon dioxide from the tissues and lungs (termed the **Haldane effect**).

As Bicarbonate

About 60% of the carbon dioxide diffuses from the tissues to red blood cells to be transported to the lungs in a chemical combination with sodium in the plasma, as sodium bicarbonate. In the red blood cells, carbon dioxide reacts with water inside the red blood cell to form carbonic acid ($\text{CO}_2 + \text{H}_2\text{O} \rightarrow \text{H}_2\text{CO}_3$). An enzyme called **carbonic anhydrase** located in red blood cells causes the reaction of water and carbon dioxide to form carbonic acid (H_2CO_3) 12,000 times faster than it would take place without the enzyme (such as in the blood plasma). As fast as carbonic acid is formed, it dissociates into a hydrogen ion (H^+) and a **bicarbonate ion** (HCO_3^-), where most of the carbon dioxide is located.

Because the H^+ ions inside the cell will increase the acidity of the venous blood, especially during exercise, which increases the proton load by a factor of about 20, the hydrogen ion concentration must be buffered. This is done by hemoglobin rather than oxyhemoglobin because Hb is a better buffer. The H^+ binds to Hb ($\text{H}^+ + \text{Hb} \rightarrow \text{HHb}$). This prevents a change in pH while the HCO_3^- diffuses into the plasma. To counteract the loss of negative charges, chloride ions (Cl^-) move from the plasma to the RBC. When oxygen dissociates from hemoglobin and diffuses into the tissues, the buffering of H^+ ions is increased along with an increase in HCO_3^- ions. The end result is an increase in the transport of carbon dioxide while the blood is kept from becoming too acidic.

As the concentration of HCO_3^- is increased in the red blood cells, it diffuses into the plasma. In the plasma, sodium chloride (NaCl) dissociates, leaving sodium to react with the HCO_3^- ions, forming NaHCO_3^- . This is the primary way in which carbon dioxide is transported to the lungs. The trade-off is that the chloride ions (Cl^-) diffuse from the plasma into the red blood cells to form **potassium chloride**. This is termed the **chloride shift**, which helps to maintain the ionic (electrical) balance between the red blood cells and the plasma. Here, it is important to point out that the reaction

moves to the right, carbon dioxide diffuses into the venous blood to be transported to the lungs. The reverse of the events occurs in the lungs. When the reaction moves to the left, carbon dioxide diffuses from the red blood cells into the alveoli.

Diffusing Capacity and Transit Time

Because the surface area for gas exchange at the alveolar–capillary interface is large in healthy subjects, diffusion and gas equilibration are not a problem. The **diffusing capacity** (transfer factor) is a test of the integrity of the alveolar–capillary membrane for gas transfer. It is performed by measuring the volume of gas that diffuses through each minute for a pressure difference of 1 mmHg. At rest, this value is about $21 \text{ mL} \cdot \text{min}^{-1} \cdot \text{mmHg}^{-1}$ and $400 \text{ mL} \cdot \text{min}^{-1} \cdot \text{mmHg}^{-1}$ for oxygen and carbon dioxide, respectively. During exercise, the oxygen and carbon dioxide values are increased to approximately $65 \text{ mL} \cdot \text{min}^{-1} \cdot \text{mmHg}^{-1}$ and $1200 \text{ mL} \cdot \text{min}^{-1} \cdot \text{mmHg}^{-1}$, respectively. Diffusing capacity is likely to be high if lung disease is not present, and if there is more surface area available for the transfer of oxygen and carbon dioxide (i.e., **ventilation–perfusion ratio**).

Whereas diffusion capacity does not limit healthy subjects, **transit time** has the potential to have a negative effect on athletic performance at very high workloads. Red blood cells spend approximately 0.75 s in the pulmonary capillaries, which is more than twice the time to oxygenate the blood (0.3 s). With exercise, the increased participation of pulmonary capillaries and their alveoli accommodates the increased cardiac output without a major increase in blood flow velocity. Thus, an optimal transit time is maintained for gas exchange. If this were not the case, blood flow through the capillaries would be increased (i.e., decreased transit time) to a point of failing to saturate the hemoglobin fully.

Ventilation–Perfusion Ratio

Ventilation and perfusion are processes that should be closely matched in the normal lung. This means the ratio would be in the vicinity of 1.0 if all parts of the lungs are equally ventilated and perfused, thus leading to the concept of **ventilation–perfusion matching** (or an optimal gas exchange). Conversely, **ventilation–perfusion mismatch** is the term used when the ventilation and the perfusion of a gas exchange are not matched. In other words, it is possible that an alveolus can be ventilated but not matched with an adequate blood flow, thus gas exchange does not take place as it should. The ventilation–perfusion ratio (or \dot{V}/\dot{Q} ratio) is used to assess the efficiency and adequacy of gas exchange (the air that reaches the lungs) and pulmonary perfusion (the blood that reaches the lungs).

The impact of gravity on pulmonary perfusion, especially at the apex of the lungs, can result in a pressure that is insufficient for developing a flow. This suggests that the variation in blood flow between the upper and lower parts of the lungs in the upright posture accounts for the whole of the ventilation–perfusion ratio inequality in the normal lungs. This is why the \dot{V}/\dot{Q} ratio is less than ideal. The apex of the lung may be highly ventilated relative to blood flow, which results in poor gas exchange. For the whole lung, the ventilation–perfusion ratio is the alveolar ventilation (\dot{V}_A) divided by the pulmonary blood flow (\dot{Q} , cardiac output), thus the following calculation: \dot{V}_A/\dot{Q} ratio. A ratio less than 1.0 represents a greater blood flow than ventilation. The average value is closer to 0.84 ($\dot{V}_A = 4.2 \text{ L} \cdot \text{min}^{-1}$; $\dot{Q} = 5 \text{ L} \cdot \text{min}^{-1}$; $\dot{V}_A \div \dot{Q} = 0.84$).

When \dot{V}_A and \dot{Q} are approximately equal, the alveolar and arterial oxygen tension (PO_2) values are about 100 mmHg. But, if the \dot{V}_A/\dot{Q} ratio is decreased either by decreasing alveolar ventilation or increasing blood flow, the alveolar oxygen tension (PAO_2) decreases toward the mixed venous oxygen tension ($\text{P}\bar{\text{V}}\text{O}_2$) value.

Diffusing capacity The rate at which a gas diffuses across the alveolar–capillary membrane per unit difference in the partial pressure of the gas across the membrane.

Ventilation–perfusion ratio The ratio of the amount of air reaching the alveoli to the amount of blood reaching the alveoli.

Transit time The time required for red cells to move through the capillary.

Ventilation–perfusion matching A state of equilibrium within the respiratory system in which the ventilatory effort matches the perfusion of blood in the capillaries.

Ventilation–perfusion mismatch An imbalance between alveolar ventilation and pulmonary capillary blood flow.

The converse is true if the \dot{V}_A/\dot{Q} ratio is increased by increasing alveolar ventilation or by decreasing the blood flow; then the P_{AO_2} increases toward the inspired oxygen tension (P_{IO_2}). The point is that the alveolar oxygen tension reflects the rate of oxygen delivered to the alveoli and the volume of oxygen transferred to the pulmonary blood and consumed by the tissues. Thus, oxygen delivery depends on both alveolar ventilation and inspired oxygen tension.

Oxygen Delivery and Utilization

Oxygen delivery The collective function of the lungs, heart, and arterial system that sustains aerobic cellular metabolism throughout the body is required to deliver oxygen.

Oxygen utilization Defined as the muscles' ability to extract (use) oxygen to sustain aerobic power within seconds to minutes of a vigorous exercise, depending on the availability of oxygen, optimal distribution of blood flow in the exercising muscles, and a limitation in the rate of oxygen extraction.

Oxygen extraction ratio Defined by the slope of the relation between oxygen transport (\dot{D}_{O_2}) and oxygen uptake (\dot{V}_{O_2}): O_2 Extraction Ratio = $(Ca_{O_2} - Cv_{O_2})/Ca_{O_2}$.

The volume of oxygen available to the tissues per minute (**oxygen delivery**; \dot{D}_{O_2}) is a product of cardiac output (\dot{Q}) and arterial content of oxygen (Ca_{O_2}). The blood flow to the tissues is directly related to the integrity of the myocardium to produce the blood flow (i.e., \dot{Q} , cardiac output = HR, heart rate \times SV, stroke volume) that is appropriate for adequate oxygenation of the tissues. The volume of oxygen consumed (i.e., **oxygen utilization**) at the tissue level is termed oxygen consumption (\dot{V}_{O_2}). It is the product of cardiac output and arteriovenous oxygen difference ($\dot{Q} \times Ca_{O_2} - Cv_{O_2}$).

At rest, given the normal hemoglobin level of 15 g per 100 mL of blood (or 150 g of Hb per liter of blood) and a normal arterial oxygen saturation the equivalent of 98% at arterial oxygen content of 20 mL of oxygen (15 g per 100 mL \times 1.34 mL of oxygen per gram of Hb = 20.1) per each 100 mL of blood (or 20 volumes percent), then, if the cardiac output (\dot{Q}) is 5.0 L \cdot min⁻¹, the calculation is $\dot{D}_{O_2} = 1.34 \times 150 \times 5.0 \times 98 \div 100 = 984.9$ mL \cdot min⁻¹ (or essentially 1000 mL \cdot min⁻¹). This is a typical resting oxygen delivery (\dot{D}_{O_2}) value for an adult subject. Note that 1.34 is used to represent the *physiologic oxygen binding capacity* rather than the *theoretical value* of 1.39 mL of oxygen binding capacity of Hb. The reason is that 4% of the binding sites cannot combine with oxygen because they are occupied by carbon monoxide and methemoglobin.

If the venous content of oxygen (Cv_{O_2}) is 15 mL of O_2 per each 100 mL of blood, the arteriovenous oxygen difference (a- v_{O_2} diff) would be 5 mL of oxygen per 100 mL of blood. Given that there is 10 \times 100 mL of blood per liter of blood, then 5 \times 10 equals 50 mL of oxygen utilized at the tissues per liter of blood. Following through with the calculation of the total body oxygen utilization steps: $\dot{V}_{O_2} = \dot{Q} \times a-v_{O_2}$ diff, therefore 250 mL \cdot min⁻¹ = 5 L \cdot min⁻¹ \times 50 mL \cdot L⁻¹. Note that the 250 mL \cdot min⁻¹ consumed at the tissues equals essentially 25% of the oxygen delivered to the tissues (\dot{D}_{O_2}). Thus, 1000 mL \cdot min⁻¹ minus 250 mL \cdot min⁻¹ equals the volume of oxygen remaining in the venous blood (i.e., 740 mL \cdot min⁻¹). This is consistent with the 75% saturation of the hemoglobin in the venous blood (Sv_{O_2}) for the body as a whole at rest.

A variation on this thinking is consistent with the concept of an **oxygen extraction ratio** (O_2ER) whereby O_2ER equals oxygen utilization, which ultimately equals \dot{V}_{O_2} . For example, $O_2ER = \dot{V}_{O_2} \div \dot{D}_{O_2} \times 100$, thus 250 mL \cdot min⁻¹ \div 1000 mL \cdot min⁻¹ \times 100 = 25%. Normal oxygen extraction ratios range from 25 to 30%. Twenty-five percent of 1000 mL \cdot min⁻¹ is 250 mL \cdot min⁻¹, or, in this case, resting \dot{V}_{O_2} . It is important that oxygen is present in sufficient amounts in the mitochondria to maintain effective concentrations of ATP in the electron transport system.

In general, \dot{V}_{O_2} reflects metabolic demands and remains relatively independent of \dot{D}_{O_2} . During heavy exercise, the increase in cellular metabolism can require 15 times the resting oxygen consumption, or 15 \times 250 mL \cdot min⁻¹ = 3750 mL \cdot min⁻¹. A \dot{Q} of 25 L \cdot min⁻¹ is required to match the \dot{V}_{O_2} requirement. A concomitant rise in \dot{D}_{O_2} is a function of the increase in \dot{Q} . The rise in \dot{D}_{O_2} is smaller in magnitude than the increase in exercise \dot{V}_{O_2} , leading to the increase in the oxygen extraction ratio as well as a progressive increase in the a- v_{O_2} diff. This improved ability to extract oxygen is

directly a function of the capacity of tissues to consume oxygen.

Study Questions

1. How does minute ventilation differ from alveolar ventilation?
2. In what way is the upper respiratory system designed to carry out its functions?
3. How does internal respiration differ from external respiration?
4. During strenuous exercise, what additional muscles help with respiration?
5. What are the static measures of lung volume and capacity?
6. What is anatomical dead space?
7. What is the difference between anatomical and physiologic dead space?
8. What is the role of myoglobin in facilitating oxygen transfer to the mitochondria during strenuous exercise?
9. When alveolar ventilation (in liters of air per minute) and alveolar capillary blood flow (in liters of blood per minute) are approximately equal. What is the effect on oxygen equilibration across the alveolar capillary membrane?
10. What is the significance of the different static lung volumes?
11. What is the Bohr effect? How is it related to blood carbon dioxide?
12. What is the Haldane effect?
13. Where and why does the chloride shift take place?
14. What determines whether oxygen and hemoglobin are combined or dissociated (separated)?
15. Why does hemoglobin combine with oxygen in the lungs and release oxygen at the tissues?
16. How would a high \dot{V}/\dot{Q} ratio affect gas exchange in the lungs?
17. Discuss the ways in which CO_2 is transported in the blood.

Suggested Readings

- Adams, G. M. (1994). *Exercise physiology laboratory manual* (2nd ed.). Dubuque, IA: Wm. C. Brown.
- Brooks, G., Fahey, T., & Baldwin, K. (2005). *Exercise physiology: Human bioenergetics and its applications* (4th ed.). New York, NY: McGraw-Hill.
- Fox, S. (2008). *Human physiology*. New York, NY: McGraw-Hill.
- Powers, S. K., & Howley, E. T. (2009). *Exercise physiology: Theory and application to fitness and performance*. (7th ed.). New York, NY: McGraw-Hill.
- Robergs, R. A., & Roberts, S. O. (1997). *Exercise physiology: Exercise, performance, and clinical applications*. New York, NY: Mosby-Year Book.
- West, J. B. (1990). *Best and Taylor's physiological basis of medical practice* (12th ed.). Baltimore, MD: Williams & Wilkins.
- Wilmore, J. H., & Costill, D. L. (2004). *Physiology of sport and exercise*. Champaign, IL: Human Kinetics.

References

- Brooks, G., Fahey, T., & Baldwin, K. (2005). *Exercise physiology: Human bioenergetics and its applications* (4th ed.). New York, NY: McGraw-Hill.
- Cottrell, G. P. (2001). *Cardiopulmonary anatomy and physiology for respiratory care practitioners*. Philadelphia, PA: F. A. Davis.
- Dempsey, I., & Fregosi, R. (1985). Adaptability of the pulmonary system to changing metabolic requirement. *American Journal of Cardiology*, 55, 59D–67D.
- Feldman, I., & Del Negro, C. (2006). Looking for inspiration: New perspectives on respiratory rhythm. *Nature Reviews*, 7, 232–242.
- Fox, S. (2008). *Human physiology*. New York, NY: McGraw-Hill.
- Guyton, A. C., & Hall, J. E. (2000). *Textbook of medical physiology*. New York, NY: W. B. Saunders.
- Leff, A. R., & Schumacker, P. T. (1993). *Respiratory physiology: Basics and applications*. Philadelphia, PA: W. B. Saunders.
- Mathews, D. K., & Fox, E. L. (1971). *The physiological basis of physical education and athletics*. Philadelphia, PA: W. B. Saunders.
- McArdle, W. D., Katch, F. I., & Katch, V. L. (2006). *Essentials of exercise physiology* (3rd ed.). Baltimore, MD: Lippincott Williams & Wilkins.
- Plowman, S. A., & Smith, D. L. (2003). *Exercise physiology for health, fitness, and performance* (2nd ed.). San Francisco, CA: Benjamin Cummings.
- Powers, S., & Beadle, R. (1985). Control of ventilation during sub-maximal exercise: A brief review. *Journal of Sports Sciences*, 3, 51–65.

- Rhoades, R., & Pflanzer, R. (1989). *Human physiology*. Philadelphia, PA: Saunders College Publishing.
- Seeley, R. R., Stephens, T. D., & Tate, P. (2003). *Anatomy and physiology* (6th ed.). Boston, MA: McGraw-Hill.
- Sherwood, L. (2007). *Human physiology: From cells to systems* (6th ed.). Belmont, CA: Thomas Brooks/Cole.
- Silverthorn, D. U. (2001). *Human physiology: An integrated approach*. (2nd ed.). Upper Saddle River, NJ: Prentice Hall.
- Slonim, N. B., & Hamilton, L. H. (1976). *Respiratory physiology* (3rd ed.). St. Louis, MO: Mosby.
- West, I., & Wagner, P. (1991). Ventilation-perfusion relationships. In R. Crystal & I. West (Eds.), *The lung: Scientific foundations*. (Vol 2, pp. 1289–1305). New York, NY: Raven Press.
- West, J. B. (1990). *Respiratory physiology: The essentials* (5th ed.). Baltimore, MD: Williams & Wilkins.
- Wilmore, J. H., & Costill, D. L. (2004). *Physiology of sport and exercise*. Champaign, IL: Human Kinetics.